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Regulating Medicines and Medical Devices

NMR in the European and US Pharmacopoeias

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Chapter 30 in *NMR in Pharmaceutical Sciences*. Edited by
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© 2015 John Wiley & Sons, Ltd. ISBN: 978-1-118-66025-6

Also published in eMagRes (online edition)

DOI: 10.1002/9780470034590.emrstm1403

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The purpose of a pharmacopoeia



- Publicly available set of quality standards for medicinal products
- Compliance is mandatory where a pharmacopoeia is referred to in drug regulations
- Monographs for individual active substances, excipients or for formulated preparations
- Supported by general notices and methods
- Product or article must comply throughout its period of use or lifetime – *if tested*



First Italian Pharmacopoeia: Ricettario Fiorentino

21 January 1499



- Giovanni Cipriani, Georgofili World, February 13 2015 and March 16 2015



- The Council of Europe's Convention on the Elaboration of a European Pharmacopoeia (1964) and amendment protocol of 1994
- 37 member states (including all those in the European Union)
- European Union Directives on human and veterinary medicines specify use of pharmacopoeias (Ph Eur or national)



- United States Pharmacopeia Convention publishes USP–NF
- combines two compendia, the United States Pharmacopeia (USP) and the National Formulary (NF)
- Recognized under the Federal Food, Drug, and Cosmetic Act.
- USP: monographs for drug substances, dosage forms and compounded preparations,
- NF: excipient monographs



- 2.2.33 – Nuclear Magnetic Resonance Spectrometry
 - General requirements
 - Nuclei (mainly ^1H and ^{13}C)
 - Quantitative and qualitative techniques
 - Fourier Transform NMR
 - Parameters to be controlled
 - Solid-state spectroscopy eg for polymorphism
- 2.2.64 - Peptide Identification by Nuclear Magnetic Resonance Spectrometry
- Technical guides



- General Chapter <761> Nuclear Magnetic Resonance Spectroscopy: qualification (installation, operational and performance), validation or verification
- General information chapter <1761> Applications of Nuclear Magnetic Resonance Spectroscopy: NMR applications
- General information chapter <1238> ‘Vaccines for Human Use – Bacterial Vaccines’: determination of depolymerized polysaccharides in protein purification section using O-acetyl groups



History of NMR in pharmacopoeias



- Mid 1970s in British Pharmacopoeia (BP) identification and quantification of the components of gentamicin (NMR spectroscopy as a general method in an Appendix)
- Quantitative determination of moisture content in Cloprostenol and Fluprostenol in the BP (Vet) 1977
- Identification tests for some corticosteroid sodium phosphates and aminoglycosidic antibiotics in the BP 1980
- Now superseded by Ph. Eur. monographs



History in Ph Eur



- Appendix on NMR Spectroscopy was first published in the second edition in 1980
- Early 1980s monograph for Gentamicin Sulfate, later replaced by liquid chromatography (LC) test



Use of NMR in monographs



- Identification using ^1H and ^{13}C NMR spectroscopy
- Assay for content or composition, frequently for oligo- or polymeric materials, using ^1H and ^{13}C NMR spectroscopy
- Related substances, including isomers, and other
- impurities using ^1H NMR spectroscopy
- Specific tests, usually to determine composition of oligo- or polymeric substances
- Relaxivity for paramagnetic lanthanide diagnostic agents



Identification and Assay by ^{13}C NMR Spectroscopy: Farmed Salmon Oil and Farmed Cod Liver Oil (Ph. Eur.)



^{13}C NMR procedure for the determination of positional distribution of ($\beta(2)$ -acyl) of fatty acids and identification of the drug substance

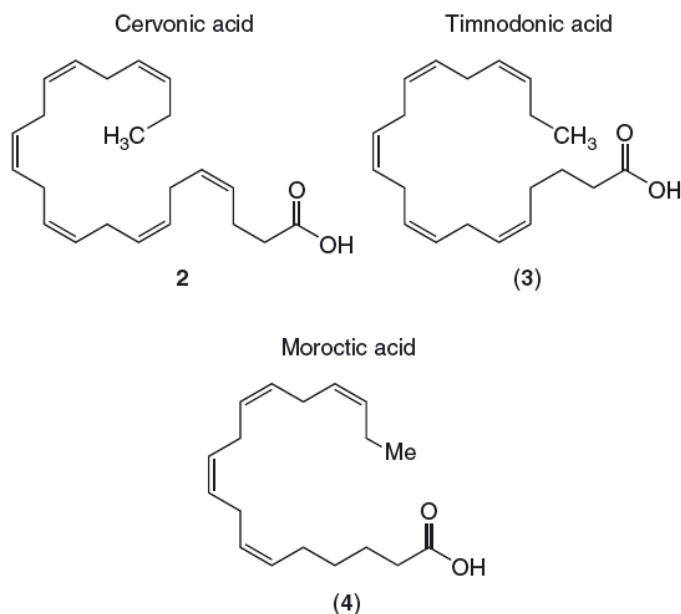


Table 30.3. ($\beta(2)$ -acyl) positional distribution in farmed salmon oil and farmed cod liver oil

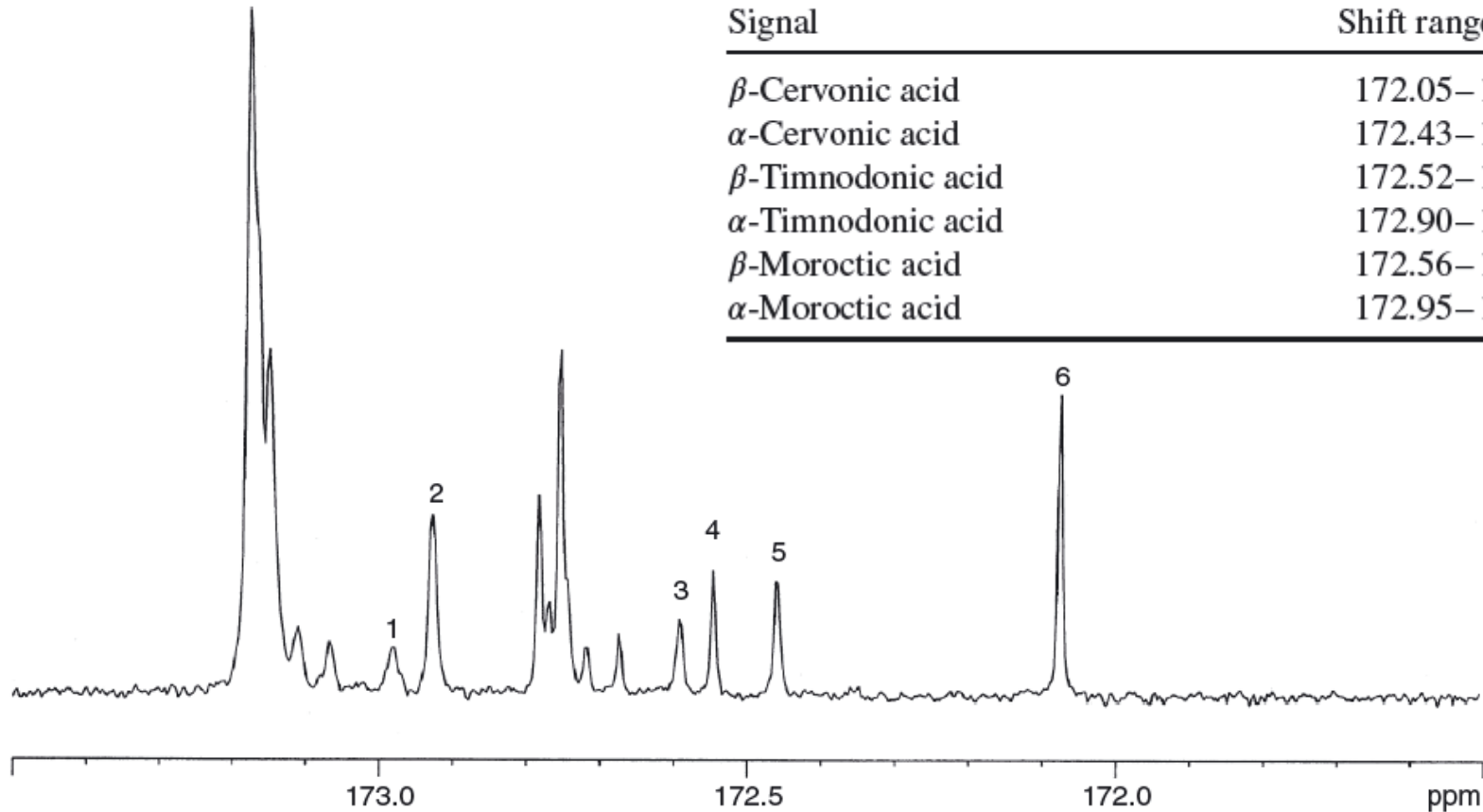
	($\beta(2)$ -acyl) positional distribution (%)	
	Farmed Salmon Oil	Farmed Cod Liver Oil
Cervonic acid	60–70	71–81
Timnodonic acid	25–35	32–40
Moroctic acid	40–55	28–38



^{13}C NMR reference spectrum for farmed salmon oil for identification

Table 30.4. ^{13}C NMR chemical shifts for fish oil omega-3 fatty acids

Signal	Shift range (ppm)
β -Cervonic acid	172.05–172.09
α -Cervonic acid	172.43–172.47
β -Timnodonic acid	172.52–172.56
α -Timnodonic acid	172.90–172.94
β -Moroctic acid	172.56–172.60
α -Moroctic acid	172.95–172.99



$$\frac{\beta}{\alpha + \beta} \times 100$$

α = peak area of α -carbonyl

β = peak area of β -carbonyl

Table 30.3. ($\beta(2)$ -acyl) positional distribution in farmed salmon oil and farmed cod liver oil

	<u>($\beta(2)$-acyl) positional distribution (%)</u>	
	Farmed Salmon Oil	Farmed Cod Liver Oil
Cervonic acid	60–70	71–81
Timnodonic acid	25–35	32–40
Moroctic acid	40–55	28–38



Relative/normalization method USP/Ph Eur



$$\alpha = (A_2/A_1) - 1$$

$$\text{Result} = 3300 \times \alpha / (33 \times \alpha + 58)$$

A1= average area of the oxypropylene methyl group doublet at about 1.08 ppm compared to the reference standard

A2 = average area of composite band in the range 3.2–3.8 ppm due to CH₂O of both the oxyethylene and oxypropylene units and the CHO groups of the oxypropylene units

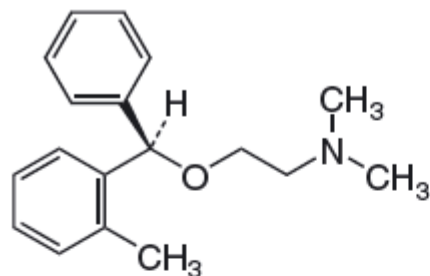


Isomeric-related Substances by ^1H NMR spectroscopy: Orphenadrine Citrate USP

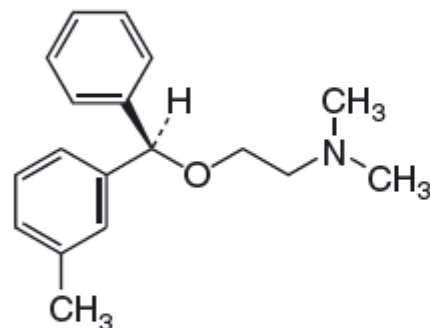
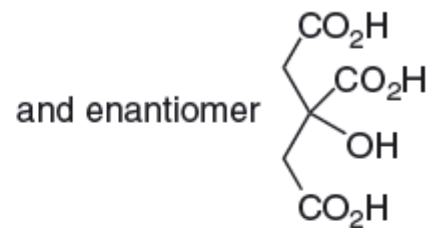
Relative method of
quantitation to limit
meta- and *para*-
isomers to 3.0%
using methine peak.

*now superseded by gas
chromatography of Ph
Eur

Orphenadrine and its impurities A and B

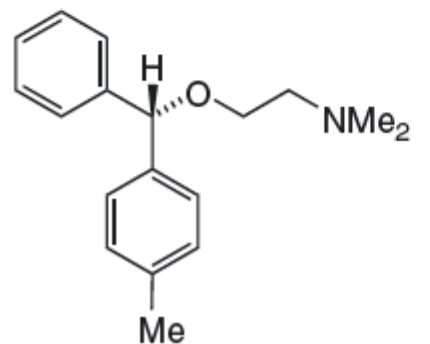


(7)



(8)

and enantiomer



(9)

and enantiomer



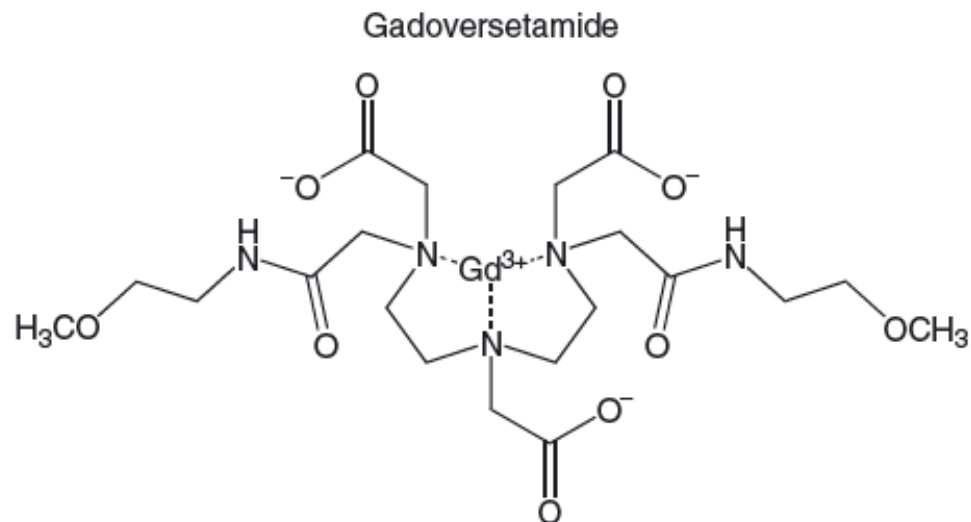
Relaxivity: Gadoversetamide Injection USP



- Relaxivity is the magnitude of a substance's capacity to enhance the relaxation rate of a nucleus, expressed in units of $s^{-1} \text{ mM}^{-1}$.
- Relaxivity of a substance is determined experimentally by measuring the spin-lattice relaxation time (T_1) of a test substance and plotting $1/T_1$ against the concentration in units of mM
- The slope of the curve is the numerical relaxivity



Relaxivity: Gadoversetamide Injection USP



- Test solutions are prepared by diluting 5.0 ml of the injection in water to give 0.504, 1.008, 2.016, and 3.024 mM, respectively
- The slope of the plotted line, the relaxivity, should be between 4.0 and 5.0 s⁻¹ mM⁻¹.



Some other examples

- Identification and Assay by ^{13}C NMR Spectroscopy: Heparin Sodium (USP and Ph. Eur.)
- Assay for Drug Substance Content by ^1H NMR Spectroscopy: Amyl Nitrite and Amyl Nitrite Inhalant USP
- Assay for Control of Functional Group Substitution by ^1H NMR Spectroscopy: Hydroxypropyl Starch Ph. Eur. And USP
- Test for Composition by ^{13}C NMR Spectroscopy: Lauromacrogol 400 Ph. Eur.
- Specific Test for Amino Acid Content by ^{13}C NMR Spectroscopy: Goserelin Acetate USP
- Impurities by ^1H NMR Spectroscopy: Medronic Acid for Radiopharmaceutical Preparations Ph. Eur.



Concluding remarks



- Pharmacopoeias provide quality standards and methods for a wide range of users and have to be generally accessible
- Not always the case with NMR techniques - sophisticated and relatively expensive equipment
- Only included in monographs when other methods not feasible or when an NMR technique has something unique to offer
- Unique capability for structural discrimination is exploited in monographs, particularly for polymeric compounds of natural or biological origin
- Characterisation of reference standards



Ricettario Fiorentino

- *“D’ingegno et di corpo destro, di buoni costumi, non avaro e fedele”*



- Giovanni Cipriani, Georgofili World, February 13 2015 and March 16 2015



- Grazie
- Quallsiasi domanda?

